

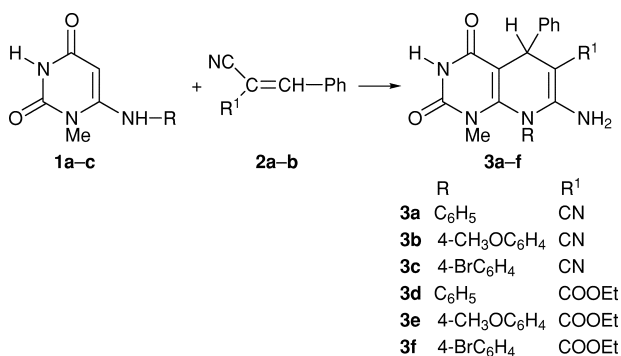
# A Facile One-pot Synthesis of Pyrido[2,3-*d*]-pyrimidines and Pyrido[2,3-*d*:6,5-*d'*]dipyrimidines†

Shaker Youssif,\* Said El-Bahaie and Esam Nabih

Department of Chemistry, Faculty of Science, University of Zagazig, Egypt

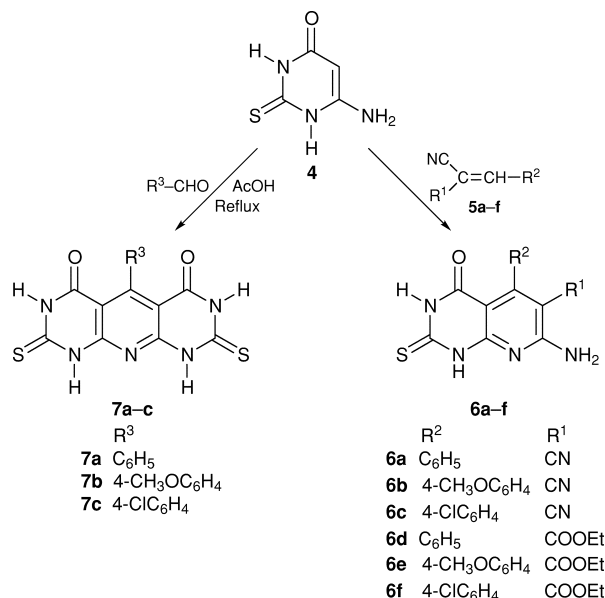
The reactions of enamionones (6-aminouracils) **1a–c** and **4** with cyano olefins **2a–b** and **5a–f** led to the formation of pyrido[2,3-*d*]pyrimidines **3a–f** and **6a–f** in good yield, while the treatment of 4-amino-2-thiouracil **4** with aromatic aldehydes afforded pyridodipyrimidines **7a–c**.

The preparation of pyrido[2,3-*d*]pyrimidines has been reported by a number of investigators and may involve suitably substituted 6-aminouracils<sup>1–4</sup> followed by annelation of the pyridine ring or cyclocondensation of 2-amino-3-cyanopyridines<sup>5–8</sup> with suitable reagents. The importance of pyrido[2,3-*d*]pyrimidines as biologically active compounds includes their use as antitumor,<sup>9</sup> antibacterial<sup>10–13</sup> and anticonvulsive<sup>14</sup> agents. This prompted us to continue our research program on the cyclization of 6-anilino-uracil derivatives.<sup>15</sup> We allowed 1-methyl-6-anilino-uracils **1** to react with reactive cyano compounds like benzylidenemalononitrile<sup>16</sup> **2a** and benzylidenecyanoacetate **2b** producing 6-cyano-5,8-dihydropyrido[2,3-*d*]pyrimidine-2,4(1*H*,3*H*)-diones **3a–c** and the corresponding ethoxycarbonyl compounds **3a–f** respectively, as shown in Scheme 1. The products were fully characterized through spectral and elemental analysis. The structure of **3a** was indicated by broad IR bands at 3460–3360 cm<sup>-1</sup> corresponding to the chelated amino group and at 2210 cm<sup>-1</sup> corresponding to CN group; the <sup>13</sup>C NMR spectrum shows 17 lines.



Scheme 1

On the other hand, the reaction of 4-amino-2-thiouracil<sup>17</sup> **4** with arylidenemalononitriles **5a–c** and arylidenecyanoacetates **5d–f** afforded the thiones **6a–c** and **6d–f** respectively, as shown in Scheme 2, fully confirmed by spectral and elemental analysis. The IR spectra of **6a** exhibited sharp bands at 3439 and 3321 cm<sup>-1</sup> (NH<sub>3</sub>), and 2215 cm<sup>-1</sup> (CN); Its NMR <sup>1</sup>H spectrum showed a singlet at δ 7.72 (NH<sub>2</sub>) and a doublet at δ 7.50–7.29 (aromatic protons). The yields of compounds **6a–f** were lower than of **3a–f**; this may be due to the higher nucleophilicity at C-5 of *N*-alkyluracils than that of uracil itself. It has been found that the intramolecular cyclization of 6-(*N*-alkylanilino)uracils with dimethylformamide (DMF)–POCl<sub>3</sub><sup>18</sup> or with *o*-haloaryl-aldehydes<sup>19</sup> in DMF or with arylaldehydes<sup>20</sup> in acetic acid afforded 5-deazaflavins. In the present work, we found that



Scheme 2

the treatment of 4-amino-2-thiouracil **4** with heating with aryl aldehydes in acetic acid under reflux afforded pyrido-dipyrimidines **7a–c** as shown in Scheme 2; 5-deazaflavins were not obtained.

The structures of compounds **7a–c** were established on the basis of satisfactory analytical and spectral data and particularly the <sup>13</sup>C NMR spectra which showed only 9 lines for **7a** and 10 lines for **7b**; the mass spectrum of **7b** gave a molecular ion peak at *m/z* 385.

## Experimental

Melting points were determined with an electrothermal Gallenkamp apparatus and are uncorrected. The <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on a Bruker AC 250 spectrometer in with (CD<sub>3</sub>)<sub>2</sub>SO as solvent and with Me<sub>4</sub>Si as internal standard. Electron impact mass spectra were recorded on a Mat 1125 70 eV spectrometer. IR spectra were recorded on a Perkin-Elmer 1430 spectrometer. The microanalyses were performed in the micro-analytical laboratory, University of Cairo, Giza, Egypt.

**General Procedure for the Synthesis of 6-Substituted 2-Amino-5-phenyl-8-aryl-5,8-dihydropyrido[2,3-*d*]pyrimidine-2,4(1*H*,3*H*)-dione 3a–f.**—A mixture of equimolar amounts of uracils **1** and the appropriate 3-phenyl-2-substituted cyano olefins **2** (2 mmol) in absolute ethanol (10 ml) in the presence of trimethylamine (5 drops) was heated at reflux for 3–6 h. The reaction mixture was concentrated and then cooled. The solid obtained was filtered off, washed with ethanol and recrystallized from DMF–EtOH (2:1). The products **3a–f** were obtained in 70–90% yield and their physical constants are as follows:

**3a:** Yield (83%); mp 302–303 °C; <sup>1</sup>H NMR δ 11.10 (s, 1H), 7.35–7.15 (m, *J<sub>n</sub>* 7, *J<sub>m</sub>* 3 Hz, 10H), 6.04 (brs, 2H), 4.66 (s, 1H), 2.78 (s, 3H); <sup>13</sup>C NMR δ 31.94 (N–C-1), 36.68 (C-5), 66.94, 102.94, 119.97, 126.68, 127.03, 127.16, 127.87, 128.24, 129.47, 141.38, 143.76, 148.81, 151.20, 155.20, 161.47; *m/z* (%) 371 (9), 307 (14), 305 (23), 304 (44), 295 (18), 294 (100), 261 (10), 251 (19), 228 (21),

\*To receive any correspondence.

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102 (13), 91 (18), 77 (30), 66 (48) (Calc. for  $C_{21}H_{17}N_5O_2$ : C, 67.91; H, 4.61; N, 18.85. Found: C, 67.88; H, 4.60; N, 18.79%).

**3b**: Yield (89%); mp 319–320 °C;  $^1H$  NMR  $\delta$  10.95 (s, 1H), 7.38–7.16 (m,  $J_m$  3,  $J_o$  8 Hz, 7H), 6.92–6.87 (d,  $J_m$  2,  $J_o$  6.8 Hz, 2H), 5.92 (s, 2H), 4.64 (s, 1H), 3.76 (s, 3H), 2.92 (s, 3H);  $^{13}C$  NMR  $\delta$  32.17 (N–C-1), 36.59 (C-5), 55.67, 66.39, 102.17, 114.94, 120.09, 126.69, 127.04, 128.28, 128.89, 133.85, 144.02, 148.95, 151.29, 155.42, 158.97, 161.49;  $m/z$  (%) 401 (11), 337 (20), 335 (63), 334 (61), 325 (16), 324 (84), 258 (71), 121 (14), 102 (12), 77 (10), 66 (100) (Calc. for  $C_{22}H_{19}N_5O_3$ : C, 65.82; H, 4.77; N, 17.44. Found: C, 65.76; H, 4.74 N, 17.38%).

**3c**: Yield (78%); mp 308–309 °C;  $^1H$  NMR  $\delta$  11.40 (s, 1H), 7.59–7.56 (d,  $J_o$  8.8 Hz, 2H), 7.34–7.19 (m,  $J_m$  2.3,  $J_o$  8.1 Hz, 7H), 6.46 (s, 2H), 4.62 (s, 1H), 2.75 (s, 3H);  $^{13}C$  NMR  $\delta$  31.81 (N–C-1), 36.40 (C-5), 66.48, 103.22, 120.09, 120.61, 126.76, 126.88, 128.31, 128.94, 132.25, 140.58, 143.35, 148.29, 151.05, 155.10, 161.44;  $m/z$  (%) 451 (8), 449 (8), 384 (9), 383 (7), 382 (8), 375 (14), 374 (97), 372 (100), 331 (19), 329 (20), 294 (27), 155 (11), 140 (7), 127 (7), 102 (17), 77 (16); (Calc. for  $C_{21}H_{16}BrN_5O_2$ : C, 56.00; H, 3.58; N, 15.54. Found: C, 55.89; H, 3.56; N, 15.37%).

**3d**: Yield (85%); mp 290–293 °C;  $^1H$  NMR  $\delta$  10.93 (s, 1H), 7.42–7.15 (m, 10H), 7.12–7.09 (d, 2H), 5.14 (s, 1H), 4.10–4.01 (q, 2H), 2.80 (s, 3H), 1.11–1.06 (t, 3H);  $^{13}C$  NMR  $\delta$  14.19 (C–CH<sub>3</sub>), 31.58 (N–C-1), 34.16 (O–C), 58.97, 86.00, 105.14, 124.78, 126.57, 127.22, 127.42, 127.78, 129.26, 141.55, 145.60, 148.52, 151.26, 156.21, 161.61, 168.25;  $m/z$  (%) 418 (18), 346 (12), 345 (41), 342 (14), 341 (100), 295 (18), 77 (48) (Calc. for  $C_{23}H_{22}N_4O_4$ : C, 66.01; H, 5.29; N, 13.38. Found: C, 65.92; H, 5.27; N, 13.21%).

**3e**: Yield (80%); mp 268–260 °C;  $^1H$  NMR  $\delta$  10.64 (brs, 1H), 7.28–7.25 (d,  $J$  7.3 Hz, 2H), 7.20–7.12 (m,  $J_m$  3,  $J_o$  6.5 Hz, 5H), 7.10–7.07 (d,  $J_o$  7,  $J_m$  2.7 Hz, 2H), 6.88–6.84 (d,  $J_m$  2,  $J_o$  6.9 Hz, 2H), 5.13 (s, 1H), 4.10–4.01 (q, 2H), 3.75 (s, 3H), 2.82 (s, 3H), 1.26–1.06 (t, 3H);  $^{13}C$  NMR  $\delta$  14.19 (C–CH<sub>3</sub>), 31.86 (N–C-1), 34.00, 55.64, 58.89, 85.24, 104.14, 114.82, 125.96, 127.21, 127.81, 128.64, 133.98, 145.93, 148.56, 151.33, 156.37, 158.75, 161.90, 168.27;  $m/z$  (%) 448 (10), 376 (10), 375 (38), 372 (20), 371 (100), 325 (13), 77 (61), 43 (12) (Calc. for  $C_{24}H_{24}N_4O_5$ : C, 64.27; H, 5.39; N, 12.49. Found: C, 64.23; H, 5.36; N, 12.35%).

**3f**: Yield (74%); mp 257–259 °C;  $^1H$  NMR  $\delta$  10.95 (s, 1H), 7.48–7.44 (d,  $J$  8.8 Hz, 2H), 7.24–7.08 (m,  $J_m$  3,  $J_o$  6.6 Hz, 9H), 5.11 (s, 1H), 4.10–4.02 (q, 2H), 2.83 (s, 3H), 1.12–1.06 (t, 3H);  $^{13}C$  NMR  $\delta$  14.19 (C–CH<sub>3</sub>), 31.40 (N–C-1), 34.23 (O–C), 59.06, 87.00, 105.96, 119.95, 126.02, 127.15, 127.85, 127.88, 132.20, 140.99, 145.24, 148.25, 151.16, 156.06, 161.88, 168.13;  $m/z$  (%) 498 (6), 497 (10), 425 (19), 424 (51), 421 (16), 420 (100), 77 (58), 43 (11) (Calc. for  $C_{23}H_{21}BrN_4O_4$ : C, 55.53; H, 4.25; N, 11.26. Found: C, 55.49; H, 4.23; N, 11.18%).

**Synthesis of 6-Substituted 7-Amino-5-aryl-2-thioxo-2,3-dihydro-pyrido[2,3-d]pyrimidin-4(1H)-one 6a–f**.—The compounds **6a–f** were prepared in 70–90% yield by the method described for the synthesis of **3a–f**.

**6a**: Yield (79%); mp > 380 °C;  $^1H$  NMR  $\delta$  12.65 (s, 1H), 12.07 (s, 1H), 7.61 (s, 2H), 7.42–7.40 (m,  $J_m$  3,  $J_o$  6.2 Hz, 3H), 7.28–7.25 (d,  $J_m$  2.3,  $J_o$  9 Hz, 2 h);  $^{13}C$  NMR  $\delta$  90.13, 100.35, 115.05, 126.83, 127.54, 128.80, 136.29, 154.30, 157.22, 158.80, 160.85, 175.96;  $m/z$  (%) 295 (100), 294 (49), 262 (12), 237 (9), 146 (28), 77 (51) (Calc. for  $C_{14}H_9N_5OS$ : C, 56.95; H, 3.07; N, 23.71. Found: C, 56.88; H, 2.98; N, 23.54%).

**6b**: Yield (72%); mp > 380 °C;  $^1H$  NMR  $\delta$  12.70 (brs, 1H), 12.07 (s, 1H), 7.68 (s, 2H), 7.24–7.21 (d,  $J_o$  8.7 Hz, 2H), 6.98–6.95 (d,  $J_o$  8.6 Hz, 2H), 3.81 (s, 3H);  $^{13}C$  NMR  $\delta$  55.02 (O–C), 78.11, 90.18, 100.31, 112.94, 115.13, 128.06, 129.32, 133.28, 154.22, 157.33, 159.44, 175.73;  $m/z$  (%) 325 (100), 324 (32), 292 (15), 267 (13), 77 (18), 66 (72) (Calc. for  $C_{15}H_{11}N_5O_2S$ : C, 55.38; H, 3.40; N, 21.53. Found: C, 55.29; H, 3.34; N, 21.37%).

**6c**: Yield (78%); mp > 360 °C;  $^1H$  NMR  $\delta$  12.70 (brs, 1H), 12.08 (s, 1H), 7.72 (s, 2H), 7.50–7.46 (d,  $J_m$  2,  $J_o$  8.4 Hz, 2H), 7.33–7.29 (d, 2H);  $^{13}C$  NMR  $\delta$  89.93, 100.28, 114.96, 127.69, 129.52, 133.22, 135.13, 154.15, 157.31, 157.47, 160.75, 175.88;  $m/z$  (%) 331 (36), 330 (32), 329 (100), 328 (45), 296 (14), 271 (11), 180 (23), 164 (9) (Calc. for  $C_{14}H_8ClN_5OS$ : C, 51.00; H, 2.44; N, 21.24. Found: C, 50.89; H, 2.41; N, 21.00%).

**6d**: Yield (86%); mp 288–289 °C;  $^1H$  NMR  $\delta$  11.59 (s, 1H), 11.51 (s, 1H), 8.06 (dd, 2H), 7.5 (m, 3H), 6.37 (s, 2H), 4.37–4.29 (q, 2H), 1.34–1.29 (t, 3H);  $^{13}C$  NMR  $\delta$  13.89 (C–CH<sub>3</sub>), 62.31 (O–CH<sub>2</sub>), 78.11, 102.28, 125.38, 129.22, 130.68, 131.28, 133.29, 138.01, 154.28, 155.01, 161.61, 174.79;  $m/z$  (%) 342 (10), 270 (11), 269 (38), 266 (20), 220 (12), 77 (15), 43 (10) (Calc. for  $C_{16}H_{14}N_4O_3S$ : C, 56.13; H, 4.12; N, 16.36. Found: C, 56.05; H, 4.09; N, 16.11%).

**6e**: Yield (75%); mp 272–274 °C;  $^1H$  NMR  $\delta$  11.42 (s, 1H), 11.27 (s, 1H), 8.08–8.04 (d,  $J_o$  8.8 Hz, 2H), 7.15–7.11 (d,  $J_o$  8.8 Hz, 2H), 6.30 (s, 2H), 4.35–4.27 (q, 2H), 3.88 (s, 3H), 1.34–1.28 (t, 3H);  $^{13}C$  NMR  $\delta$  13.95 (C–CH<sub>3</sub>), 55.73 (O–CH<sub>3</sub>), 62.01 (O–CH<sub>2</sub>), 78.27, 98.84, 114.94, 116.05, 124.00, 131.21, 133.40, 154.30, 161.56, 162.32, 163.56, 174.71;  $m/z$  (%) 372 (20), 300 (12), 299 (31), 296 (19), 219 (100), 173 (52), 77 (71), 43 (12) (Calc. for  $C_{17}H_{16}N_4O_4S$ : C, 54.83; H, 4.33; N, 15.04. Found: C, 54.68; H, 4.31; N, 14.90%).

**6f**: Yield (79%); mp 345–346 °C;  $^1H$  NMR  $\delta$  11.35 (s, 1H), 10.99 (brs, 1H), 7.45–7.41 (d,  $J_o$  8.3 Hz, 2H), 7.27–7.24 (d,  $J_o$  8.3 Hz, 2H), 5.09 (s, 2H), 4.15–4.06 (q, 2H), 1.20–1.14 (t, 3H);  $^{13}C$  NMR  $\delta$  13.55 (C–CH<sub>3</sub>), 62.12 (O–CH<sub>2</sub>), 72.20, 93.66, 116.62, 127.45, 129.48, 132.65, 136.54, 155.16, 159.61, 162.71, 163.51, 174.78;  $m/z$  (%) 378 (30), 377 (22), 376 (100), 304 (39), 294 (15), 261 (10), 288 (18), 77 (59) (Calc. for  $C_{16}H_{13}ClN_4O_3S$ : C, 51.00; H, 3.47; N, 14.87. Found: C, 50.94; H, 3.44; N, 14.80%).

**General Procedure for the Synthesis of 5-Substituted 1,3,8,9-Tetrahydro-2,8-dithioxopyrido[2,3-d:6,5-d']pyrimidine-4,6(1H,7H)-diones 7a–c**.—A solution of 6-amino-2-thiouracil **4** (2.32 mmol) in glacial acetic acid (15 ml) and 0.5 equiv. of the appropriate aromatic aldehyde was heated under reflux for 4 h. The reaction mixture was diluted with water, then allowed to cool to room temperature. The crude product was collected and recrystallized from a suitable solvent. The physical constants are as follows:

**7a**: Yield (71%); mp 278–280 °C (from ethanol);  $^1H$  NMR  $\delta$  12.02 (s, 2H), 11.81 (brs, 2H), 7.25–7.06 (m, 5H);  $^{13}C$  NMR  $\delta$  72.19, 90.18, 125.22, 126.38, 127.75, 137.85, 153.38, 162.93, 172.72;  $m/z$  (%) 355 (18), 295 (15), 270 (43), 109 (100), 77 (19), 68 (71) (Calc. for  $C_{13}H_9N_5O_2S_2$ : C, 50.71; H, 2.55; N, 19.71. Found: C, 50.68; H, 2.53; N, 19.55%).

**7b**: Yield (73%); mp 292–294 °C (from ethanol–DMF 2:1);  $^1H$  NMR  $\delta$  11.94 (s, 2H), 11.75 (brs, 2H), 6.97–6.76 (dd, 4H), 3.70 (s, 3H);  $^{13}C$  NMR  $\delta$  54.85 (O–CH<sub>3</sub>), 72.19, 90.45, 113.18, 127.43, 129.47, 153.32, 157.03, 162.89, 172.67;  $m/z$  (%) 385 (14), 263 (41), 260 (87), 143 (100), 121 (74), 115 (44), 68 (86), 43 (78) (Calc. for  $C_{16}H_{11}N_5O_2S_2$ : C, 49.87; H, 2.87; N, 18.17. Found: C, 49.80; H, 2.83; N, 18.01%).

**7c**: Yield (81%); mp 334–336 °C (from DMF–ethanol 5:1);  $^1H$  NMR  $\delta$  11.89 (s, 2H), 11.70 (s, 2H), 7.26–7.08 (dd, 4H);  $^{13}C$  NMR  $\delta$  72.23, 89.99, 127.59, 128.46, 129.86, 137.17, 153.37, 162.91, 172.91;  $m/z$  (%) 391 (26), 389 (48), 329 (16), 304 (45), 264 (66), 205 (23), 143 (100) (Calc. for  $C_{13}H_8ClN_5O_2S_2$ : C, 46.22; H, 2.06; N, 17.97. Found: C, 46.18; H, 2.00; N, 17.86%).

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## References

- H. Ogura and M. Sakaguchi, *Chem. Lett.*, 1972, 657.
- G. L. Anderson, *J. Heterocycl. Chem.*, 1985, **22**, 1469.
- T. Itoh, I. Fujii, Y. Tomii, I. Ishikawa, H. Ogura, Y. Mizuno and N. Kawahara, *J. Heterocycl. Chem.*, 1987, **24**, 1453.
- P. Bhuyan, R. C. Boruah and J. S. Sandhu, *J. Org. Chem.*, 1990, **55**, 568.
- S. Youssif and M. Assy, *J. Chem. Res.*, 1996, (S) 442; (M) 2546.
- W. S. Emerson and T. M. Patric, *J. Org. Chem.*, 1949, **14**, 795; F. D. Popp and A. Catala, *J. Org. Chem.*, 1961, **26**, 2738.
- E. C. Taylor and C. C. Cheng, *J. Org. Chem.*, 1960, **25**, 148.
- F. Yoneda, Y. Sakuma, S. Mazumoto and R. Ito, *J. Chem. Soc., Perkin Trans. I*, 1976, 1805.
- E. M. Grivsky, S. Lee, C. W. Sigel, D. S. Duch and C. A. Nichol, *J. Med. Chem.*, 1980, **23**, 327.
- J. Matsumoto and S. Minami, *J. Med. Chem.*, 1975, **18**, 74.
- N. Suzuki, *Chem. Pharm. Bull.*, 1980, **28**, 761.
- V. Oakes and H. N. Rydon, *J. Chem. Soc.*, 1956, 4433.
- J. I. DeGraw, R. L. Kisliuk, Y. Gaumont and C. M. Baugh, *J. Med. Chem.*, 1974, **17**, 470.
- E. Kretzchmar, *Pharmazie*, 1980, **35**, 253.
- P. J. Vanderhost and C. S. Hammiton, *J. Am. Chem. Soc.*, 1953, **75**, 656.
- R. K. Robins and G. H. Hitchings, *J. Am. Chem. Soc.*, 1955, **77**, 2256.
- K. Tserng and L. Bauer, *J. Heterocycl. Chem.*, 1972, **9**, 1433.
- B. Stanovnik and M. Tisler, *Synthesis*, 1972, **6**, 308.
- T. Nagamatsu, Y. Hashiguchi, M. Higuchi and F. Yoneda, *J. Chem. Soc., Chem. Commun.*, 1982, 1085.
- F. Yoneda, *Lectures in Heterocycl. Chem.*, 1980, **5**, S73.